TopSpec

2019 2020 2021 2022 R R WP4 – Development & Application of CE D4.1 – Prototype CED gun at KI T4.1 – design 1KeV pulsed e source T4.2 – construction T4.3 – interfacing CED gun with omni T4.4 – testing CED T4.5 – optimizing software & hardware D4.2 – Optimized CE guns at KI/IP T4.6 – analysis of mAbs M18 M23 M32 M36 **M1** M13

Jan 2019



NIST Atomic Spectra Database - Ionization Energies Data

At. Num.	Sp. Name.	Ion Charge	El. name	lsoel. Seq.	Ground Shells	Ground Level	Ionized Level	lonization Energy (eV)	Uncertainty (eV)	References
6	CI	0	Carbon	С	1s ² 2s ² 2p ²	³ P ₀	2s ² 2p ² P° _{1/2}	11.2602880	0.0000011	L20057
6	CII	+1	Carbon	В	1 <i>s</i> ²2 <i>s</i> ²2 <i>p</i>	² P° _{1/2}	2 <i>s</i> ² ¹ S ₀	24.383154	0.000016	c190
6	C III	+2	Carbon	Be	1 <i>s</i> ² 2 <i>s</i> ²	¹ S ₀	2s ² S _{1/2}	47.88778	0.00025	L876c191
6	CIV	+3	Carbon	Li	1 <i>s</i> ²2 <i>s</i>	² S _{1/2}	1 <i>s</i> ^{2 1} S ₀	64.49352	0.00019	L11667
6	CV	+4	Carbon	He	1 <i>s</i> ²	¹ S ₀	1s ² S _{1/2}	[392.090515]	0.000025	L10054
6	C VI	+5	Carbon	Н	1s	² S _{1/2}		(489.993194)	0.000007	L7188

At. Num.	Sp. Name.	Ion Charge	El. name	Isoel. Seq.	Ground Shells	Ground Level	Ionized Level	lonization Energy (eV)	Uncertainty (eV)	References
7	NI	0	Nitrogen	N	$1s^22s^22p^3$	⁴ S° _{3/2}	2p ^{2 3} P ₀	14.53413	0.00004	L1411
7	NII	+1	Nitrogen	С	$1s^2 2s^2 2p^2$	³ P ₀	2p ² P° _{1/2}	[29.60125]	0.00009	L11770
7	N III	+2	Nitrogen	В	1s ² 2s ² 2p	² P° _{1/2}	2 <i>s</i> ² ¹ S ₀	[47.4453]	0.0025	L11770
7	NIV	+3	Nitrogen	Be	1 <i>s</i> ² 2 <i>s</i> ²	¹ S ₀	2s ² S _{1/2}	77.4735	0.0004	L7288,L876
7	NV	+4	Nitrogen	Li	1s ² 2s	² S _{1/2}	1 <i>s</i> ² ¹ S ₀	97.8901	0.0004	L4829
7	N VI	+5	Nitrogen	He	1s ²	¹ S ₀	1 <i>s</i> ² S _{1/2}	[552.06732]	0.00004	L10054
7	N VII	+6	Nitrogen	Н	1s	² S _{1/2}		(667.046116)	0.000013	L7188
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At. Num.	Sp. Name.	ion Charge	EI. name	isoei. Seq.	Ground Shells	Ground Level	ionized Level	(eV)	(eV)	References
At. Num.	OI	lon Charge	EI. name Oxygen	o	1s ² 2s ² 2p ⁴	³ P ₂	2p ^{3 4} S° _{3/2}	(eV)	(eV) 0.000007	L74,L3760
At. Num. 8 8	01 01	0 +1	Cxygen Oxygen	O N	$1s^22s^22p^4$ $1s^22s^22p^3$	³ P ₂ ⁴ S° _{3/2}	$2p^{3} {}^{4}S^{\circ}{}_{3/2}$ $2p^{2} {}^{3}P_{0}$	(eV) 13.618055 35.12112	(eV) 0.000007 0.00006	L74,L3760
At. Num. 8 8 8	Sp. Name. 01 011 011	0 +1 +2	EI. name Oxygen Oxygen Oxygen	o N C	1s ² 2s ² 2p ⁴ 1s ² 2s ² 2p ³ 1s ² 2s ² 2p ²	³ P ₂ ⁴ S° _{3/2} ³ P ₀	$2p^{3} {}^{4}S^{\circ}{}_{3/2}$ $2p^{2} {}^{3}P_{0}$ $2p {}^{2}P^{\circ}{}_{1/2}$	(eV) 13.618055 35.12112 [54.93554]	0.000007 0.000007 0.00006 0.00012	L74,L3760 L11267,L10621 L11770
At. Num. 8 8 8 8	Sp. Name. 01 0II 0III 0IV	0 +1 +2 +3	EI. name Oxygen Oxygen Oxygen Oxygen	O N C B	Shells 1s ² 2s ² 2p ⁴ 1s ² 2s ² 2p ³ 1s ² 2s ² 2p ² 1s ² 2s ² 2p	³ P ₂ ⁴ S° _{3/2} ³ P ₀ ² P° _{1/2}	$\begin{array}{c} 2p^{3} {}^{4}S^{\circ}{}_{3/2} \\ 2p^{2} {}^{3}P_{0} \\ 2p {}^{2}P^{\circ}{}_{1/2} \\ 2s^{2} {}^{1}S_{0} \end{array}$	(eV) 13.618055 35.12112 [54.93554] 77.41350	(eV) 0.000007 0.00006 0.00012 0.00025	L74,L3760 L11267,L10621 L11770 L648
At. Num. 8 8 8 8 8 8	Sp. Name. 01 011 011 011 011 011 011 011 011 011 011 011 011 011 011 011 011 011 011 011	0 +1 +2 +3 +4	EI. name Oxygen Oxygen Oxygen Oxygen Oxygen	O N C B Be	Shells 1s ² 2s ² 2p ⁴ 1s ² 2s ² 2p ³ 1s ² 2s ² 2p ² 1s ² 2s ² 2p 1s ² 2s ² 2p	³ P ₂ ⁴ S° _{3/2} ³ P ₀ ² P° _{1/2} ¹ S ₀	$\begin{array}{c} 2p^{3} \ ^{4}\text{S}^{\circ}_{\ 3/2} \\ 2p^{2} \ ^{3}\text{P}_{0} \\ 2p \ ^{2}\text{P}^{\circ}_{1/2} \\ 2s^{2} \ ^{1}\text{S}_{0} \\ 2s \ ^{2}\text{S}_{1/2} \end{array}$	(eV) 13.618055 35.12112 [54.93554] 77.41350 113.8990	(eV) 0.000007 0.00006 0.00012 0.00025 0.0005	L74,L3760 L11267,L10621 L11770 L648 L7288
At. Num. 8 8 8 8 8 8 8 8	Sp. Name. 01 011 0111 0111 01V 0V 0V1	0 +1 +2 +3 +4 +5	EI. name Oxygen Oxygen Oxygen Oxygen Oxygen Oxygen	O N C B Be Li	Ground Shells 1s ² 2s ² 2p ⁴ 1s ² 2s ² 2p ² 1s ² 2s ² 2p 1s ² 2s ² 2p 1s ² 2s ² 1s ² 2s ²	Ground Level ³ P ₂ ⁴ S° _{3/2} ³ P ₀ ² P° _{1/2} ¹ S ₀ ² S _{1/2}	$\begin{array}{c} 2p^{3} {}^{4} {\mathbb{S}^{\circ}}_{3/2} \\ 2p^{2} {}^{3} {\mathbb{P}_{0}} \\ 2p {}^{2} {\mathbb{P}^{\circ}}_{1/2} \\ 2s^{2} {}^{1} {\mathbb{S}_{0}} \\ 2s {}^{2} {\mathbb{S}_{1/2}} \\ 1s^{2} {}^{1} {\mathbb{S}_{0}} \end{array}$	(eV) 13.618055 35.12112 [54.93554] 77.41350 113.8990 [138.1189]	(eV) 0.000007 0.00006 0.00012 0.00025 0.0005 0.0021	L74,L3760 L11267,L10621 L11770 L648 L7288 L4713
At. Num. 8 8 8 8 8 8 8 8 8 8	Sp. Name. 01 011 0111 0111 0112 012 012 013 014 015 010 011 012 012 013 014 015 </td <td>0 +1 +2 +3 +4 +5 +6</td> <td>EI. name Oxygen Oxygen Oxygen Oxygen Oxygen Oxygen</td> <td>O N C B Be Li He</td> <td>Science 1s²2s²2p⁴ 1s²2s²2p³ 1s²2s²2p² 1s²2s²2p 1s²2s² 1s²2s² 1s²2s 1s²2s</td> <td>Ground Level ³P₂ ⁴S°_{3/2} ³P₀ ²P°_{1/2} ¹S₀ ¹S₀</td> <td>$\begin{array}{c} 2p^{3} {}^{4}S^{\circ}{}_{3/2} \\ 2p^{2} {}^{3}P_{0} \\ 2p {}^{2}P^{\circ}{}_{1/2} \\ 2s^{2} {}^{1}S_{0} \\ 2s {}^{2}S_{1/2} \\ 1s^{2} {}^{1}S_{0} \\ \end{array}$</td> <td>Ionization Energy (eV) 13.618055 35.12112 [54.93554] 77.41350 113.8990 [138.1189] [739.32682]</td> <td>Oncertainty (eV) 0.000007 0.00006 0.00012 0.00025 0.0005 0.0021 0.00006</td> <td>L74,L3760 L11267,L10621 L11770 L648 L7288 L4713</td>	0 +1 +2 +3 +4 +5 +6	EI. name Oxygen Oxygen Oxygen Oxygen Oxygen Oxygen	O N C B Be Li He	Science 1s ² 2s ² 2p ⁴ 1s ² 2s ² 2p ³ 1s ² 2s ² 2p ² 1s ² 2s ² 2p 1s ² 2s ² 1s ² 2s ² 1s ² 2s 1s ² 2s	Ground Level ³ P ₂ ⁴ S° _{3/2} ³ P ₀ ² P° _{1/2} ¹ S ₀ ¹ S ₀	$\begin{array}{c} 2p^{3} {}^{4}S^{\circ}{}_{3/2} \\ 2p^{2} {}^{3}P_{0} \\ 2p {}^{2}P^{\circ}{}_{1/2} \\ 2s^{2} {}^{1}S_{0} \\ 2s {}^{2}S_{1/2} \\ 1s^{2} {}^{1}S_{0} \\ \end{array}$	Ionization Energy (eV) 13.618055 35.12112 [54.93554] 77.41350 113.8990 [138.1189] [739.32682]	Oncertainty (eV) 0.000007 0.00006 0.00012 0.00025 0.0005 0.0021 0.00006	L74,L3760 L11267,L10621 L11770 L648 L7288 L4713

NIST Energy Levels and Wavelengths Bibliographic Reference # 10054

Theoretical Energies for the n=1 and 2 States of the Helium Isoelectronic Sequence up to Z=100,

G. W. F. Drake, Can. J. Phys. 66, 586–611 (1988) DOI:10.1139/p88-100

Prior Art



X-ray emission cross sections and fluorescence yields for light atoms and molecules by electron impact, Tawara H *et al*, Physica 63 (1973) 351-367

Cross Sections for Inner-Shell Ionization by Electron Impact, Llovet X et al J Phys Chem Ref Data 43, 013102 (2014); doi: 10.1063/1.4832851









Ta disc 1.6mm diam / 0.1mm thick heating current 6A

T4.2 Construction 1KeV pulsed electron source







Ta disc 1.6mm diam / 0.1mm thick heating current 6A





T4.4 Testing CE gun with Omni



T4.4 Testing CE gun with Omni



T4.4 Testing CE gun with Omni

Single ionization of $[M+7H]^{7+}$ $[M+7H]^{7+} + e_f^- \longrightarrow [M+7H]^{8+\bullet} + e_f^- + e_s^ [M+7H]^{7+} + e_f^- \longrightarrow [M+7H-CO_2]^{8+\bullet} + e_f^- + e_s^-$

weighted relative intensity = relative intensity × correction factor



Electron Induced Dissociation of ubiquitin has been performed at the two maxima observed in the ionization efficiency curves and at higher e energy (35eV, 350eV & 800eV)

Differences in EID performed at different energies are observed !



 $[M+8H]^{8+} + e_f \longrightarrow [M+8H]^{9+\bullet} + e_f + e_s \longrightarrow \text{fragments}$

